## **Rayat Shikshan Sanstha's** Rajarshi Chhatarapati Shahu College, Kolhapur B.Sc.Part-II Semester-III (New CBCS) **Preliminary Examination Oct.2019 PHYSICS Paper-V** DSC-C1 THERMAL PHYSICS AND STATISTICAL MECHANICS – I

Day & Date: Thursday 10/10/2019	Time : <b>12</b> :	00 pm to 2:00pm	Maximum Marks: 50	
<b>N B</b> 1 All questions are computed	sorv			
2 Figures to the right indic	ate the full mar	ks		
2. Use of coloulotors/locarithmic tobles is allowed				
4. Drew past diagrams wherever passagery				
4. Draw near diagrams wherever necessary.				
Q.1 Select the correct alternative from the following [10]				
A) The temperature interval between the ice point and stream point is				
1) range of thermometer	11)fundamenta	al interval		
iii) basic interval iv)normal interval				
B) Coefficient of diffusion in gas corresponds to transfer of of gas.				
i)momentum ii) ene	ergy	iii) mass	iv) entropy	
C) The average kinetic energy of a gas molecule is at absolute temperature T is proportional to				
i) 1/T	ii) T	iii) $T^2$ iv	) √T	
D) Claussius formula for mean free path of gas molecule is				
$i)\frac{1}{\pi\sigma^2 n}$ $ii)\frac{3}{4}\frac{1}{\pi\sigma^2 n}$	$iii)\frac{1}{\sqrt{2}}$	$\overline{2}\frac{1}{\pi\sigma^2 n}$ ii	$\frac{4}{3}\frac{1}{\pi\sigma^2 n}$	
E) Thermo e.m.f. produced in a thermocouple is of the order of				
i) microvolt ii) milivolt	iii) vo	lt i	/) kilovolt	
F) Zeroth law of thermodynamics leads to the definition of the term				
i) temperature ii)pres	ssure	iii) volume	iv) entropy	
G) is adiabatic relation	n			
i) PV = constant ii) PV	$\sigma^{\gamma} = \text{constant}$	iii) $P^{\gamma}V = consta$	nt iv) $PV^{\gamma-1} = constant$	
H) Heat conduction through a body is example of process.				
i) reversible ii) irre	eversible	iii) isothermal	iv) adiabatic	
I) is reversible process.				

i) Carnot's heat engine	ii) Free expansion of gas
,	,

iii) Heat conduction iv) Rubbing of stones

J) Heat engine converts heat into ------

i) light energy	ii) mechanical work
iii)electrical energy	iv) potential energy

## Q. 2 Attempt ANY TWO of the following

A) Explain transport of momentum in gases. Obtain expression for coefficient of viscosity of the gas.

B) Explain Carnot's ideal heat engine. Obtain expression for efficiency of Carnot's heat engine

working between the temperatures  $T_1$  and  $T_2$ .

C) Explain construction and working of platinum resistance thermometer.

## Q. 3 Attempt ANY FOUR of the following

[20]

[20]

A) State properties of mercury suitable for its use in thermometer.

B) Define mean free path and obtain expression for mean free path using mutual collision cross section method.

C) Explain thermodynamic equilibriums of system.

D) Obtain an expression for work done in an isothermal process.

E) Give physical significance of entropy.

F) Obtain expression for work done during an adiabatic change.