Rayat Shikshan Sanstha's

Rajarshi Chhatarapati Shahu College, Kolhapur

Department of Physics

Question Bank

Paper XIV- DSE-F2 Solid State Physics Class: B.Sc. III Teacher's name: Dr. Archana R. Patil Unit I- Chapter I- Crystal Structure • Multiple Choice Questions (Correct answer is shown in red color) The number of point group and space lattice structure in three dimensions are.... 1 a) 7,14 b) 32,14 c) 4,7 d) 14,32 2 The Miller index for the plane which does not cut the crystallographic axes ∞ (b a) 0 b) 1 c) 1 3 The close packed structure are a) SC and FCC b) BCC and FCC c) HCP and FCC d) BCC and HCP 4 The interplanar distance for plane (221) in case of cubic crystal lattice is.... a) $\frac{a}{5}$ c) $\frac{a}{2}$ d) $\frac{a}{6}$ b) $\frac{a}{0}$ 5 Primitive unit cell contains.... number of atoms c) 3 a) 1 b) 2 d) 4 The packing fraction for FCC crystal structure 6 b) $\frac{\sqrt{3\pi}}{2}$ d) $\frac{\pi}{\sqrt{2}}$ a) $\frac{\pi}{6}$ c) $\frac{\pi}{3\sqrt{2}}$ 7 The atomic radius of HCP lattice is c) $\frac{\sqrt{3}}{4}a$ d) $\frac{a}{2}$ b) $\frac{a}{2\sqrt{2}}$ a) $\frac{a}{2\sqrt{2}}$ 8 The coordination number for SC lattice a) 8 c) 12 d) 10 b) 6 9 The packing fraction for SC crystal structure

	a) 0.52	b) 0.68	c) 0.74	d) 0.34
10	The atomic radius of B	CC lattice is		
	a) $\frac{a}{2}$	b) $\frac{a}{2\sqrt{2}}$	c) $\frac{\sqrt{3}}{4}a$	d) $\frac{a}{\sqrt{2}}$
11	The coordination number	er for FCC lattice		
	a) 6	b) 8	c) 12	d) 10
12	The packing fraction for	r BCC crystal struc	ture	
	a) 0.52	b) 0.68	c) 0.74	d) 0.34
13	The atomic radius of sin	nple cubic lattice is		
	a) $\frac{a}{2}$	b) $\frac{a}{2\sqrt{2}}$	c) $\frac{\sqrt{3}}{4}a$	d) $\frac{a}{\sqrt{2}}$
14	The coordination number	er for HCP lattice		
	a) 6	b) 8	c) 12	d) 10
15	The packing fraction for	HCP crystal struct	ture	
	a) 0.52	b) 0.68	c) 0.34	d) 0.74
16	The atomic radius of FC	CC lattice is		
	a) $\frac{a}{2}$	b) $\frac{a}{2\sqrt{2}}$	c) $\frac{\sqrt{3}}{4}a$	d) $\frac{a}{\sqrt{2}}$
17	The coordination number	er for BCC lattice		
	a) 6	b) 8	c) 12	d) 10
18	The number of lattice p	oints per unit cell f	or non primitive ce	ell is
	a) equal to one		b) greater than o	ne
	c) less than one		d) equal to zero	
19	The Miller index for the	plane parallel to c	rystallographic axe	S
	a) 0	b) 1	c) $\overline{1}$	(b) (c)
20	The sequence of atoms	in hexagonal close	packed crystal stru	ucture is
	a) ABCABC	b) ABAB	c) random	d) ABCDABCD
21	The Miller index for the	plane which cuts t	he X-axis at $\frac{1}{3}\overline{a}$	is
	a) 3	b) 1	c) 3	d) $\overline{1}$

22	If $\frac{360^{\circ}}{n}$ gives n-fold through the angle of		then for diad axis tl	ne cube must be rotated
	a) 60 ⁰	b) 90°	c) 120 ⁰	d) 180 ⁰
23	The Miller indices for t	he plane which cut	the X-axis half wa	y of the unit cell are given
	by			
	a) (020)	b) (200)	c) (002)	d) $\left(\frac{1}{2}00\right)$
24	Which of the following	g has a HCP crystal	structure?	
	a) W	b) Mo	c) Cr	d) Zr
25	Amorphous solids have	structure.		
	a) Regular	b) Linear	c) Irregular	d) Dendritic
26	Which of the following	is a property of no	n-metallic crystals	?
	a) Highly ductile		b) Low electrical	conductivity
	c) Less brittle		d) FCC structure	
27	Which of the following	g is not an amorpho	ous material?	
	a) Glass	b) Plastics	c) Lead	d) Rubbers
28	Bravais lattice consists	of space latti	ces in three dimens	sions.
	a) 07	b) 13	c) 05	d) 14
29	Bravais lattice consists	of space lattic	ces in two dimensio	ons.
	a) 07	b) 13	c) 05	d) 14
30	The axial relationship of	f a monoclinic crys	stal system is given	as
	a) $a = b = c$	b) $a = b \neq c$	c) $a \neq b = c$	d) $a \neq b \neq c$
31	The interracial angles of	of a hexagonal crys	tal system are give	n by
	a) $\alpha = \beta = \gamma = 90o$		b) $\alpha = \beta = 90^{\circ}, \gamma$	
	c) $\alpha = \beta = \gamma \neq 90^{\circ}$		d) $\alpha \neq \beta \neq \gamma \neq 90$	0
32	The axial relationship o			
	a) $a = b = c$,	c) $a \neq b = c$	
33	The interracial angles o	f a triclinic crystal		
	a) $\alpha = \beta = \gamma = 900$		b) $\alpha = \beta = 90^{\circ}, \gamma$	
	c) $\alpha = \beta = \gamma \neq 90^{\circ}$		d) $\alpha \neq \beta \neq \gamma \neq 90$	U
34	Repeatable entity of a c	rystal structure is k	known as	

	a) Crystal	b) Lattice	c) Unit cell	d) Miller indices	
35	Every point of space lattice has surroundings.				
	a) Atoms	b) element	c) Lattice	d) Identical	
36	is a system of not	ation of planes wit	hin a crystal of spa	ce lattice	
	a) Space lattice	b) Crystalline	c) Miller	d) identical	
			indices		
37	Atomic packing fraction	n is maximum for			
	a) plastic	b) SC	c) BCC	d) FCC	
38	If $\frac{360^{O}}{n}$ gives n-fold a	axis of symmetry t	hen for triad axis th	e cube must be rotated	
	n gives in iola i	unis of symmetry t	non for triad axis ti	le cube must be fotuted	
	through the angle of				
	a) 60 ⁰	b) 90 ⁰	c) 120 ⁰	c) 180 ⁰	
39	The interracial angles of	f a Hexagonal crys	tal system are give	n by	
	a) $\alpha = \beta = \gamma = 90o$		b) $\alpha = \beta = 90^{\circ}, \gamma$	$= 120^{\circ}$	
	c) $\alpha = \beta = \gamma \neq 90^{\circ}$		d) $\alpha \neq \beta \neq \gamma \neq 90$	o	
40	If $\frac{360^{\circ}}{n}$ gives n-fold a	axis of symmetry t	hen for tetrad axis	the cube must be rotated	
	through the angle of				
	a) 60 ⁰	b) 90 ⁰	c) 120 ⁰	c) 180 ⁰	
41	Co-ordination no. & Ate	omic packing fract	ion of HCP structu	re is same as that of	
	the				
	a) SC	b) FCC	c) BCC	d) Rubber	
42	The angle between [111] and [11–2] direct	tions in a cubic cry	stal is (in degrees)	
	a) 0	b) 45	c) 90	d) 120	
43	Atomic packing factor is				
	a) Distance between two adjacent atoms				
	b) Projected area fraction of atoms on a plane				
	c) Volume fraction of atoms in cell				
	d) Distance between two	0 1			
44	Which unit cell has eight atoms located in the corners, has sides that are all the same				
	length, and has angles o	f only 90°?			

	a) SC	b) FCC	c) BCC	d) triclinic
45	Intercepts of a plane in c indices are,	rystal is given by a	, b/2, 3c in a simj	ple cubic unit cell, Miller
	a) (1 3 2)	b) (2 6 1)	c) (1 2 3)	d) (361)
46	The sequence of atoms	in FCC crystal struc	cture is	
	a) ABCABC		b) ABAB	
	c) random		d) ABCDABCI	D
47	A Cube has elements	of symmetry		
	a) 13	b) 01	c) 09	d) 23
48	A Cube has rotation a	axes of symmetry		
	a) 13	b) 01	c) 09	d) 23
49	A Cube has planes of	symmetry		
	a) 13	b) 01	c) 09	d) 23
50	$\frac{c}{a}$ ratio in HCP crysta	l structure is		
	a) $\sqrt{\frac{8}{3}}$	b) $\sqrt{\frac{3}{8}}$	c) $\sqrt{\frac{4}{3}}$	d) $\sqrt{\frac{3}{2}}$
~				

• Short Answer Questions

- 1. Explain the concept of reciprocal lattice.
- 2. Give construction of two dimensional reciprocal lattice.
- 3. State and explain the properties of reciprocal lattice.
- 4. Show that the volume of unit cell of reciprocal lattice is inversely proportional to the volume of the unit cell in the direct lattice.
- 5. Show that reciprocal lattice to bcc lattice is fcc lattice.
- 6. Find the reciprocal lattice to fcc lattice.
- 7. Derive Bragg's law for X-ray diffraction
- 8. What is Ewald's construction? Derive Bragg's law in reciprocal lattice.
- 9. Obtain the vector form of Bragg's law using the concept of reciprocal lattice.
- 10. What are Brillouin zones? Discuss the construction of first two Brillouin zones for a square lattice.
- 11. Explain analysis of cubic crystal by powder method.

12. Explain diffraction of X-rays by crystal.

• Long Answer Questions

- 1. Describe Laue's method of X-ray diffraction.
- 2. Describe Rotating Crystal method of X-ray diffraction.
- 3. Describe powder method of X-ray diffraction.
- 4. What is reciprocal lattice? Derive relations for primitive translation vectors of the reciprocal lattice in terms of those of the direct lattice.

Unit II- Chapter I- Magnetic Properties of Matter

- Multiple Choice Questions (Correct answer is shown in red color)
- 1. Magnetic susceptibility χ is.....
- A) dipole moment per unit volume
- B) torque per unit area
- C) magnetization for unit magnetic field intensity
- D) none of these
- 2. One Bohr magneton is....

A)
$$9.27 \times 10^{-24} amp.m^2$$
 B) $2.27 \times 10^{-24} amp.m^2$

C) $6.67 \times 10^{-24} amp.m^2$ D) $9.27 \times 10^{-9} amp.m^2$

3. Magnetic susceptibility χ of a material is given by.....

A)
$$\chi = \begin{pmatrix} \mu & -1 \\ r & \end{pmatrix}$$
 B) $\chi = \frac{M}{H}$ C) $\chi = \begin{pmatrix} \mu - \mu \\ \frac{0}{\mu} \\ o \end{pmatrix}$ D) All

4. The magnetic materials in which permanent magnetic dipoles are already aligned due to bonding forces are known as.....

- A) paramagnetic materials B) ferromagnetic materials
- C) diamagnetic materials D) antiferromagnetic materials

5. In ferromagnetic materia	l, susceptibility is			
A)very large and negative	A)very large and negative			
B)very small and negative				
C) very large and positive				
D)very small and positive				
6. Which of the following	material does not h	ave permanent magnetic di	pole	
A) paramagnetic		B) diamagnetic		
C) ferrimagnetic		D) antiferromagnetic		
7. Diamagnetic material pos	ssesses			
A) induced dipole moment				
B) permanent magnetic dip	oles			
C) no permanent magnetic of	lipoles			
D) none of these				
8. The susceptibility of diam	nagnetic material is	s about		
A) 10 ⁻⁶	в) 10 ⁷	c) 10 ⁵	D) 10 ⁻⁵	
9. The Ferromagnetic Curie temperature of iron				
A) 922 K) 922 K B) 631 K C) 1043K D) 1428 K			
10.Curie -Weiss law is				
A) $\chi = \frac{C}{T}$	B) $\chi = \frac{C}{\theta}$	C) $\chi = \frac{C}{\theta - T}$	D) $\chi = \frac{C}{T - \theta}$	

11. At Curie temperature, the spontaneous magnetization for ferromagnetic material is......

A) ∞ B) 0 C) 1 D) -1

12. Each ferromagnetic material has characteristic temperature above which its properties are vitally different from those below it. This temperature is called.....

A) demagnetization temperature

B) Curie temperature

C) Faraday's temperature

D) transition temperature

13. The paramagnetic susceptibility decreases with.....

A) increasing temperature	B) decreasing temperature
C) constant temperature	D) none of these
14. Above Curie temperature ferromagnetic s	substance becomes
A) diamagnetic	B) paramagnetic

C) ferromagnetic D) antiferromagnetic

15. The temperature at which domain structure collapses is called as..

A) Curie temperature	Neel temperature
C)Weiss temperature	transition temperature

16. The interaction between the neighbouring dipoles is negligible in case of....

A) paramagnetic materials B) diamagnetic materials

C)ferromagnetic materials D)antiferromagnetic materials

17. According to classical Langevin theory of diamagnetism in presence of magnetic field frequency of revolution of electron changes by factor....

A)
$$\pm \frac{eB}{2m}$$
 B) $\pm \frac{eB}{m}$ C) $\pm \frac{eh}{4\pi m}$ D) $\pm \frac{eh}{4m}$

18. Susceptibility of of material is independent of the temperature....

A) diamagnetic

B) paramagnetic

C) ferromagnetic D) ferrimagnetic

19. Curie law for paramagnetic material is..

20. Saturation magnetization in paramagnetic materials is observed at....

- A) high temperature and high magnetic field
- B) at high temperature and low magnetic field
- C) low temperature and low magnetic field

D)low temperature and high magnetic field

- 21. The effective number of bar magnet on each P_{eff} is.....
- A) $g\sqrt{(J+1)}$ B) $g\sqrt{J(J-1)}$ C) $g\sqrt{J(J+1)}$ D) gJ

22.Quantum theory of paramagnetism approaches to the classical Langevin theory of paramagnetism when.....

- A) $J \to 0$ B) $J \to \infty$ C) $J = -\frac{1}{2}$ D) $J = \frac{1}{2}$
- 23. The Curie law of paramagnetism holds good for.
- A) $\mu B >> kT$ B) $\mu B = kT$ C) $\mu B = \frac{1}{kT}$ D) $\mu B << kT$
- 24. Saturation magnetization in paramagnetism is Ms =
- A) $N\mu$ B) $\frac{N}{\mu}$ C) $N\mu L(x)$ D) $N^2\mu$

25Materials have large and positive value of susceptibility			
B) diamagnetic	B) paramagnetic		
C) ferromagnetic	D) antiferromagnetic		
26. The lagging intensity of magnetisation b	behind the magnetising field is called as		
A) hysteresis	B)Spontaneous magnetization		
C) saturation magnetization	D) wall displacement		
27. The value of magnetisation which rema	ins even after magnetising field is reduced to zero		
A) hysteresis	B) retentivity		
C) coercivity	D) saturation magnetization		
28. Energy loss during hystersis is the area of	of		
A) $\chi - T$ B) $B - H$	C) $M - B$ D) $\chi - H$		
29. The magnetic field B_E is proportional			
29. The magnetic field B_E is proportional			
29.The magnetic field B_E is proportionalA) Magnetization of domain	B) Applied magnetic field		
	B) Applied magnetic fieldD) area off domain		
A) Magnetization of domain	D) area off domain		
A) Magnetization of domainC) magnetic induction	D) area off domain		
 A) Magnetization of domain C) magnetic induction 30. The susceptibility of diamagnetic material 	D) area off domain als is to the atomic number.		
 A) Magnetization of domain C) magnetic induction 30. The susceptibility of diamagnetic materi A) inversely proportional 	D) area off domainals is to the atomic number.B) directly proportional		
 A) Magnetization of domain C) magnetic induction 30. The susceptibility of diamagnetic materi A) inversely proportional C) equal 	 D) area off domain als is to the atomic number. B) directly proportional D) independent 		
 A) Magnetization of domain C) magnetic induction 30. The susceptibility of diamagnetic materia A) inversely proportional C) equal Short Answer Questions 	 D) area off domain als is to the atomic number. B) directly proportional D) independent 		
 A) Magnetization of domain C) magnetic induction 30. The susceptibility of diamagnetic materia A) inversely proportional C) equal Short Answer Questions 1. State and explain Curie law in paramage 	 D) area off domain als is to the atomic number. B) directly proportional D) independent gnetism. gnetic domains formed? 		
 A) Magnetization of domain C) magnetic induction 30. The susceptibility of diamagnetic materia A) inversely proportional C) equal Short Answer Questions 1. State and explain Curie law in paramage 2. Explain how and why are the ferromage 3. Draw a typical B-H curve and described 	 D) area off domain als is to the atomic number. B) directly proportional D) independent gnetism. gnetic domains formed? 		
 A) Magnetization of domain C) magnetic induction 30. The susceptibility of diamagnetic materia A) inversely proportional C) equal Short Answer Questions 1. State and explain Curie law in paramage 2. Explain how and why are the ferromage 3. Draw a typical B-H curve and described 	 D) area off domain als is to the atomic number. B) directly proportional D) independent gnetism. gnetic domains formed? the different magnetization processes. material. What is retaintivity and coercivity? he area of B-H curve. 		

- 7. Explain the following terms briefly. (a) hysteresis, (b) coercivity, (c) remanence
- 8. Distinguish between diamagnetic, paramagnetic and ferromagnetic materials.
- 9. 8. Derive Curie Weiss law.

• Long Answer Questions

- Explain classical Langevin's theory of diamagnetism and obtain an expression for 1. diamagnetic susceptibility.
- 2. Explain classical Langevin's theory of paramagnetism. Obtain an expression for paramagnetic susceptibility.
- 3 Obtain an expression for diamagnetic susceptibility using the Langevin's theory. 6. Given an account of quantum theory of paramagnetism and discuss low and high temperature cases.
- 4 Give an account of Weiss theory of ferromagnetism. Discuss the temperature variation of saturation magnetization.
- 5 Explain the cause of hysteresis phenomenon in the ferromagnetic material. What does area of B-H loop signify?

Unit II- Chapter II- Band Theory of Solids

• Multiple Choice Questions (Correct answer is shown in red color)

- 1. The density of electron states is proportional to
- D) $E^{-l/2}$ B)¹/_E C) $E^{1/_2}$ A) E

2. In Kronig-Penny model period of one dimensional periodic potential is....

A) a	B) a+b	C) a-b	D) b
11) u	Djulo	C) u 0	\mathbf{D}

3. According to Kronig-Penny model width of allowed energy band with increase inenergy.

D)either increase or decrease

A) increases	B)decreases	

C)remains constant

4. In Kronig-Penny model if $P \rightarrow 0$, then it leads to...

- A) tight binding model B) intermediate model
- C) free electron model D) can't say anything

5. The first Brillouin zone lies between the values of k =

A) $-\frac{\pi}{a}to + \frac{\pi}{a}$ B) $0 to + \frac{\pi}{a}$ C) $-\frac{\pi}{a}to 0$ D) $-\frac{\pi}{2a}to + \frac{\pi}{2a}$

6. The velocity of an electron in one dimensional periodic potential is v.

A) $\frac{dE}{dK}$ B) $\frac{l}{\hbar} \frac{dE}{dK}$ C) $\frac{1}{\hbar} \frac{d^2 E}{dK^2}$ D) $\frac{d^2 E}{dK^2}$

7. The effective mass of an electron is $m^*=$

A) $\frac{d^2 E}{dk^2}$ B) $\frac{\frac{d^2 E}{dK^2}}{\hbar^2}$ C) $\frac{l}{\hbar} \frac{dE}{dK}$ D) $\frac{\hbar^2}{\frac{d^2 E}{dK^2}}$

8. In lower half of energy band of E-k curve the effective mass of an electron is

B) infinite C) negative A) zero D) positive 9. Band gap energy of silicon is.....eV A) 1.12 B) 0.72 C) 7 D) 0.5 10. Band gap energy of germanium is..... eV A) 1.12 **B**) 0.72 C) 7 D) 0.5 11.In semiconductors forbidden energy gap Eg is of the order of ev A) 0 C) 7 **B**) 1 D) ∞ 12. The conduction band of insulators is..... B) completely filled A) completely empty C) constant temperature D) none of these 13. The conductivity of the semiconductor...... with increase in temperature. A) decreases B) increases C) remains constant D) either increase or decrease 14. In Hall effect if the current is flowing due to motion of holes, then Hall coefficient is.....

A) positive	B) negative

C) either positive or negative D) zero

15. The Hall coefficient of the n-type semiconductor material is given as R_H

A)
$$-\left[\frac{3\pi}{8} \cdot \frac{1}{\rho}\right]$$
 B) $+\left[\frac{3\pi}{8} \cdot \frac{1}{\rho}\right]$ C) $-\frac{1}{\rho}$ D) $+\frac{1}{\rho}$

16. In Hall effect, if the current flowing through the material is in x-direction, applied magnetic field is in z-direction then Hall voltage will be produced in....

A)x-direction	B) y-direction
C) z-direction	D) x and y direction

17. The effective number of free electrons that is $N_{eff} = 0$ for....

A)metals	B) metals and semiconductors
C) semiconductors	D) insulators

• Short Answer Questions

- 1. Explain the concept of density of state.
- 2. State Bloch function.
- 3. Write a note on Effective mass of an electron.
- 4. Define valence band, conduction band and forbidden energy gap.

• Long Answer Questions

- Discuss the formation of allowed and forbidden energy bands on the basis of Kronig-Penny model.
- 2. Obtain an expression for velocity of an electron as predicted by band theory. Hence explain variation of velocity of electrons with wave vector.
- 3. Show that effective mass of an electron is inversely proportional to $\frac{d^2E}{dK^2}$. Explain its significance.
- 4. Obtain an expression for effective number of free electrons in an energy band
- 5. Define Hall Effect. Obtain an expression for Hall voltage and Hall coefficient.
- 6. Explain variation of effective mass of an electron with a wave vector.
- 7. Explain how energy gap is formed between allowed energy bands.
- 8. Distinguish between metal, semiconductor and insulator on the basis of their energy band structure.